

CHARACTERIZATION OF SILICA/SILVER-BASED ANTIBACTERIAL LEATHER

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ABSTRACT. The hydrophilic character of vegetable tanned leather is potentially a medium for bacterial growth. A treatment using an antibacterial agent applies to prevent bacterial growth on it. The vegetable tanned leather in this study was obtained from the tanning of goat skin using mimosa as a tanning agent. The utilization of silica from volcanic ash modified with silver was employed in this study as an antibacterial agent. The functional groups of materials were analysed using FTIR spectrophotometer and evaluated. The results of thermal studies using TG/DTA show that vegetable tanned leather coated with silica (leather@SiO₂) and vegetable tanned leather coated with silica/silver (leather@SiO₂/Ag) are thermally stable materials. The inhibition zones of *Staphylococcus aureus* for leather@SiO₂ and leather@SiO₂/Ag were larger compared to vegetable tanned leather with inhibition area of 21.25 ± 0.50 mm, 24.80 ± 1.64 mm and 11.40 ± 0.55 mm, respectively. This confirmed the effectiveness of utilization of silica-based volcanic ash and silver as an antibacterial agent.

KEY WORDS: leather, silica, silver, antibacterial agent

CARACTERIZAREA PIELII CU PROPRIETĂȚI ANTIBACTERIENE TRATATE CU SILICE/ARGINT

REZUMAT. Având în vedere caracterul său hidrofil, pielea tăbăcită vegetal poate fi un mediu propice pentru creșterea bacteriilor. Se poate aplica un tratament care utilizează un agent antibacterian pentru a preveni creșterea bacteriilor pe piele. Pielea tăbăcită vegetal din acest studiu a fost obținută prin tăbăcirea pielii de capră folosind mimosa ca agent de tăbăcire. Silicea din cenușa vulcanică modificată cu argint a fost utilizată în acest studiu ca agent antibacterian. Grupările funcționale ale materialelor au fost analizate folosind spectrofotometrul FTIR și apoi evaluate. Rezultatele studiilor termice efectuate folosind TG/DTA arată că pielea tăbăcită vegetal acoperită cu silice (leather@SiO₂) și pielea tăbăcită vegetal acoperită cu silice/argint (leather@SiO₂/Ag) sunt materiale stabile din punct de vedere termic. Zonele de inhibare ale *Staphylococcus aureus* pentru leather@SiO₂ și leather@SiO₂/Ag au fost mai mari comparativ cu pielea tăbăcită vegetal cu suprafață de inhibare de 21,25 ± 0,50 mm, 24,80 ± 1,64 mm și, respectiv, 11,40 ± 0,55 mm. Acest lucru a confirmat eficacitatea utilizării cenușii vulcanice pe bază de silice și a argintului ca agent antibacterian.

CUVINTE CHEIE: piele, silice, argint, agent antibacterian

CARACTÉRISATION DU CUIR ANTIBACTÉRIEN TRAITÉ AVEC DE SILICE/D'ARGENT

RÉSUMÉ. En raison de sa nature hydrophile, le cuir au tannage végétal peut être un environnement approprié pour la croissance des bactéries. Un traitement peut être appliqué qui utilise un agent antibactérien pour empêcher la croissance bactérienne sur la peau. Le cuir au tannage végétal dans cette étude a été obtenu en tannant la peau de chèvre en utilisant du mimosa comme agent de tannage. La silice provenant des cendres volcaniques modifiée par l'argent a été utilisée dans cette étude comme agent antibactérien. Les groupes fonctionnels des matériaux ont été analysés à l'aide du spectrophotomètre FTIR puis évalués. Les résultats des études thermiques utilisant TG/DTA montrent que le cuir tanné végétal traité avec de silice (leather@SiO₂) et le cuir tanné végétal traité avec de silice/argent (leather@SiO₂/Ag) sont des matériaux thermiquement stables. Les zones d'inhibition de *Staphylococcus aureus* pour leather@SiO₂ et leather@SiO₂/Ag étaient plus grandes par rapport au cuir au tannage végétal avec une surface inhibitrice de 21,25 ± 0,50 mm, 24,80 ± 1,64 mm et 11,40 ± 0,55 mm. Cela a confirmé l'efficacité de l'utilisation de cendres volcaniques à base de silice et d'argent comme agent antibactérien.

MOTS CLÉS : cuir, silice, argent, agent antibactérien

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INTRODUCTION

The processes of leather tanning play an important role in obtaining the character of the product, especially the tanning process. In general, the skin is tanned using mineral or vegetable tanning agents. Mineral tanning agents that are usually used are chrome compounds derived from Cr(III) salts. This species of Cr(III) may be oxidized forming Cr(VI) which is known to be a hazardous material to the environment. Therefore, the tanning of skin using chrome compounds needs to be minimized.

The alternative tanning process may be conducted using vegetable tanning agents. These vegetable tanning agents are extracted from plants that are known as renewable natural resources. Vegetable tanning agents generally contain a high number of hydroxyl (–OH) groups, increasing the hydrophilicity and the humidity of the tanned leather. As a result, vegetable tanned skin is potentially a good bacterial growth medium. Bacteria that grow on vegetable tanned skin can lead to a disease for consumers. Hence, it is necessary to make an effort to obtain antibacterial tanned skin.

Antibacterial agents such as pentachlorophenol, polyhalogenated phenolic compounds, dimethylfumarate, quaternary ammonium salts, silylquaternary compounds, etc., have been used in leather industry. However, some of them are harmful to human health and the environment [1-3]. Therefore, studies to find alternative antibacterial agents have been widely developed.

Silica is one of the antimicrobial agents which has been studied in several fields [4-8]. It can be utilized as antibacterial particles or as delivery systems for antibacterial agents [6, 8]. Various sources of silica have been used as silica precursors, such as tetraethoxysilane (TEOS) [9-11] and water glass [12]. Volcanic ash is solid waste from an eruption that has the potential to be used as a raw material for the preparation of a silica precursor due to its high silica content. Furthermore, volcanic ash of Mount Merapi has been known to contain 51.31% silica [13]. However, the utilization of silica as an antibacterial agent has a limited effect against bacteria [6]. One attempt to enhance the antibacterial activity of silica is by modification via metal particles [5, 11, 14]. The

previous studies found that silver (Ag) can be used as an antibacterial agent in leather [15-17]. Silver has attracted attention due to its excellent antibacterial activity and non-toxic character [6, 15, 16]. Moreover, it has been reported that the attachment of silica-silver for various fields giving good antibacterial effect [5, 9, 10, 12, 16]. Thus, this study focuses on applying the silica-based coating modified with silver on leather.

EXPERIMENTAL

Materials and Methods

Materials

The materials used for the vegetable tanning process were three pieces of pickled goat skins and mimosa as a tanning agent. Silica was extracted from ash of Merapi Volcano, Yogyakarta, Indonesia using sodium hydroxide. Other chemicals used included sodium formate, naphthalene sulphonate (Coralon OT produced by STAHL), replacement syntan (Tanicor PWB produced by STAHL), sulphited oil (Derminol OCS produced by STAHL) and silver nitrate (Merck).

Methods

Synthesis of Silica from Volcanic Ash

The synthesis of silica was employed as conducted by Sudjarwo and Bee (2017) [20]. Merapi Volcano ash was mixed with sodium hydroxide by the ratio (w/w) of 1:1. The mixture was heated to 900°C in the furnace for 3 (three) hours. Afterward, the solid product was dissolved in water, after which 2 M hydrochloric acid was added. This mixture was stirred for a day then filtered. The filtrate containing silica was used as a material for coating leather.

Coating Leather with Silver Modified Silica

The pickled goat skins were processed by the vegetable tanning method for insole shoes (Table 1). Two obtained pieces of leather were coated using the impregnation method [21]. One leather was coated by silica while the other one was coated by silica and AgNO₃ solution. These coated leathers were dried and stored in closed plastic bags before the treatment for antibacterial activity assays.

Table 1: The stages of leather preparation process

Stage Process	Chemical Material (%)
pH Adjustment	200% water 0.1% Sodium Formate (60 minutes)
Tanning	200% water 5% Dispersing Agent (Coralon OT) (30 minutes) 10% Mimosa Sul (60 minutes) 10% Mimosa (60 minutes) 10% Mimosa (60 minutes) 5% Tanicor PWB (60 minutes)
Post Tanning	
Fatliquoring	4% Derminol OCS (45 minutes)
Coating processes	1% Silica Solution (2 hours) 1% Formic Acid (45 minutes)
Fixation	1% AgNO ₃ solution

Identification of Functional Groups with Fourier Transform Infrared (FTIR) Spectrophotometry

FTIR spectrophotometry analysis was performed using Shimadzu FTIR Prestige 21. An amount of 2 mg of sample was homogenized with 200 mg of KBr powder. It was made into a pellet using 2000 psi in pressure. The absorbance of the sample was recorded at a wavenumber range of 400-4000 cm⁻¹.

Thermal Analysis with Thermo-Gravimetry and Differential Thermal Analysis (TG/DTA)

TG/DTA analysis was conducted using Perkin-Elmer under a nitrogen gas flow with a heating rate of 10.00°C/min within the temperature range of 30.00-750.00°C. About 5 mg of sample was prepared in aluminium sample holder then heated according to the standard operating above.

Antibacterial Activity Assays

The bacteria inoculum was prepared by aseptically transferring isolated colonies (*Staphylococcus aureus*) to the nutrient broth, then incubated during 24 h at 37 ± 1°C. The inoculum was diluted to 0.5 McFarland turbidity standards (corresponding to a concentration of 1.5 – 3.0 × 10⁸ CFU/mL). For antibacterial activity evaluation purposes, a coated leather sample (2 cm × 2 cm) was placed on a Petri dish containing 0.5 mL of the working bacterial dilution in 20 mL nutrient agar solution. Afterward, the Petri dishes were incubated for 24 h at 37 ± 1°C.

The evaluation of the antibacterial activity was conducted on the presence of an inhibition growth zone around the edges of the tested leather sample.

RESULTS AND DISCUSSIONS

The tanning process in this study was employed using goat skin as raw material and mimosa as a vegetable tanning agent. Mimosa contains polyphenols that can interact with collagen of goat skin via hydrogen and covalent bonding. The abundance of hydroxyl groups on the vegetable tanned surface makes it a good bacterial growth medium. Therefore, the vegetable tanned leather was coated using silica and silver to give the antibacterial effect on the leather.

The tanning process leads to an increase in weight of leather (Table 2). This indicates the formation of the bond between the tanning agent and goat skin [22]. As can be seen in Table 2, leather@SiO₂ and leather@SiO₂/Ag show a higher percentage of weight increase compared to vegetable tanned leather. This can be explained by the modification of leather with silica and Ag that generated the interaction between modifier and leather.

Table 2: Percentage of weight increase after process

Sample	Weight of pickled skin (g)	Weight of leather (g)	Percentage of increase (%)
Vegetable tanned leather	398.8	458.5	14.52
Leather@SiO ₂	279.6	321.3	14.90
Leather@SiO ₂ /Ag	379.8	437.5	15.19

The silica was obtained from the Merapi volcanic ash which is abundant and unused material. The treatment of volcanic ash was conducted using sodium hydroxide and hydrochloric acid resulting in sodium silicate. Sodium silicate can react with the hydroxyl groups of vegetable tanned leather forming

other hydroxyl groups on the coated vegetable tanned leather. The modification of the coating layer was done using silver. The positive charges of silver interacted with negative charges of protonated oxygen resulting in electrostatic interaction. The illustrations of coated vegetable tanned leather surface are shown in Figure 1.

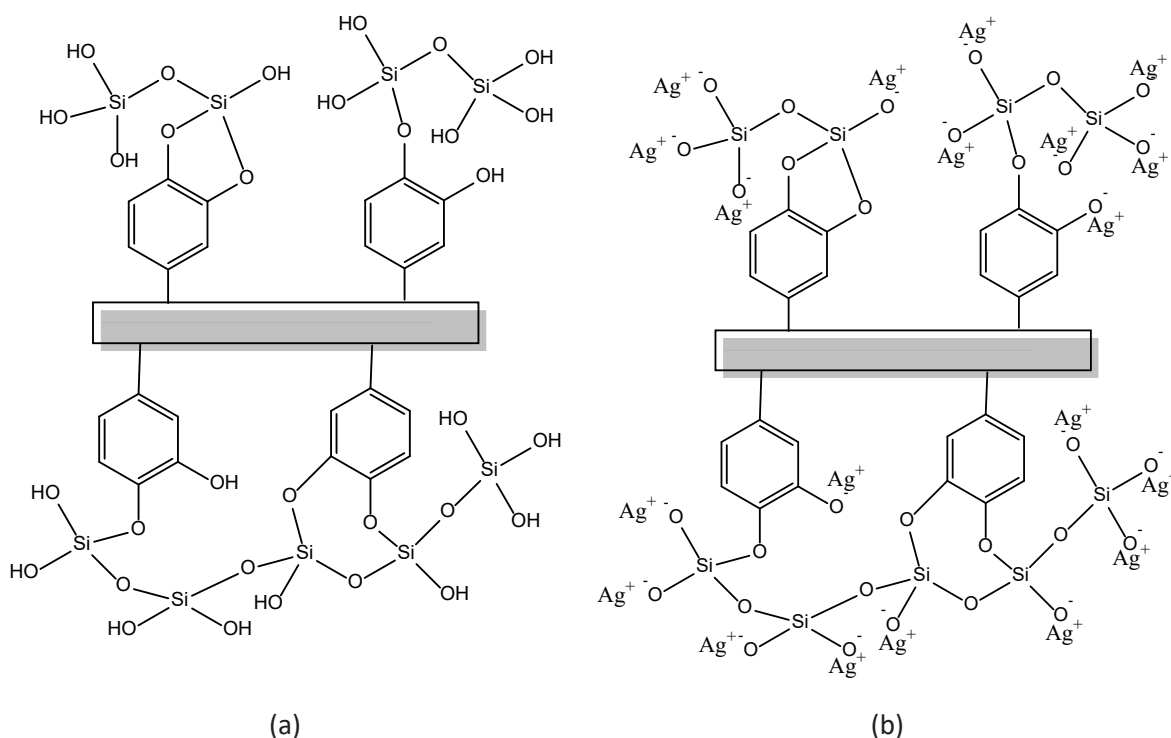


Figure 1. Interaction model of vegetable tanned leather coated with: (a) silica-based volcanic ash and (b) silica-based volcanic ash/Ag

Identification of Functional Groups of Leather

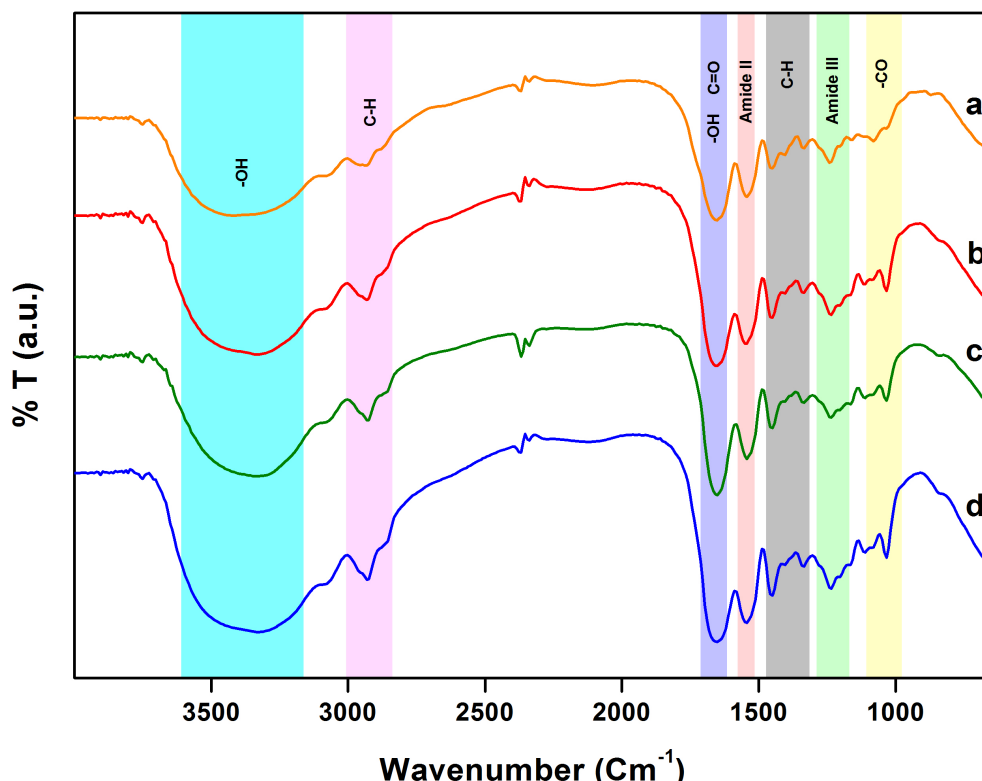


Figure 2. FTIR spectra of (a) pickled goat skin, (b) vegetable tanned leather, (c) leather@SiO₂, and (d) leather@SiO₂/Ag

The amide groups of leather were identified in several wavelength regions. Amide I band resulted in absorbance at 1651 cm⁻¹ corresponding to C=O stretching vibration of pure collagen [23]. This band overlapped with -OH bending vibration. The absorption band at 1543 cm⁻¹ appeared from the amide II band [23, 24]. Amide III band is shown between 1234-1242 cm⁻¹ attributing to symmetrical stretching of carboxylate groups [17, 24]. The absorption bands at 1335; 1450 and around 2924-2932 cm⁻¹ appeared as C-H bending, C-H wagging, and C-H asymmetric stretching vibration respectively, verifying the presence of C-H bonding in collagen [25-27].

It is shown in the FTIR spectra that the coating with silica does not change the absorption bands of tanned leather. This can be explained by the silica characteristic bands that commonly appeared around 1034-1080 cm⁻¹ and 3400 cm⁻¹ overlapped with -CO and -OH stretching bands of tannin [7, 12, 28-31]. However, the attachment of Ag on leather is not

revealed in the FTIR spectra. This phenomenon suggests that no bond occurs between Ag and the modified leather, indicating that Ag and modified leather interact electrostatically. It also confirmed that the modification of leather using Ag does not affect the structure of modified leather [24]. The FTIR spectra revealed no difference in characteristic absorption band between one another. It implies that the modifiers' penetration occurs almost uniformly from the flesh to the grain of the leather.

Stability at Thermal Decomposition

The TG analysis curves of samples are shown in Figure 3. The thermal degradation of samples showed three stages of weight loss over a temperature range of 30-750°C. The first stages occurred at 40-90°C due to the water content of samples [32, 33]. The second stage ranges between 150-500°C and was identified as the decomposition of the organic moiety [34]. The last stage was observed above 500°C, caused by the weight loss of carbonized residues [33].

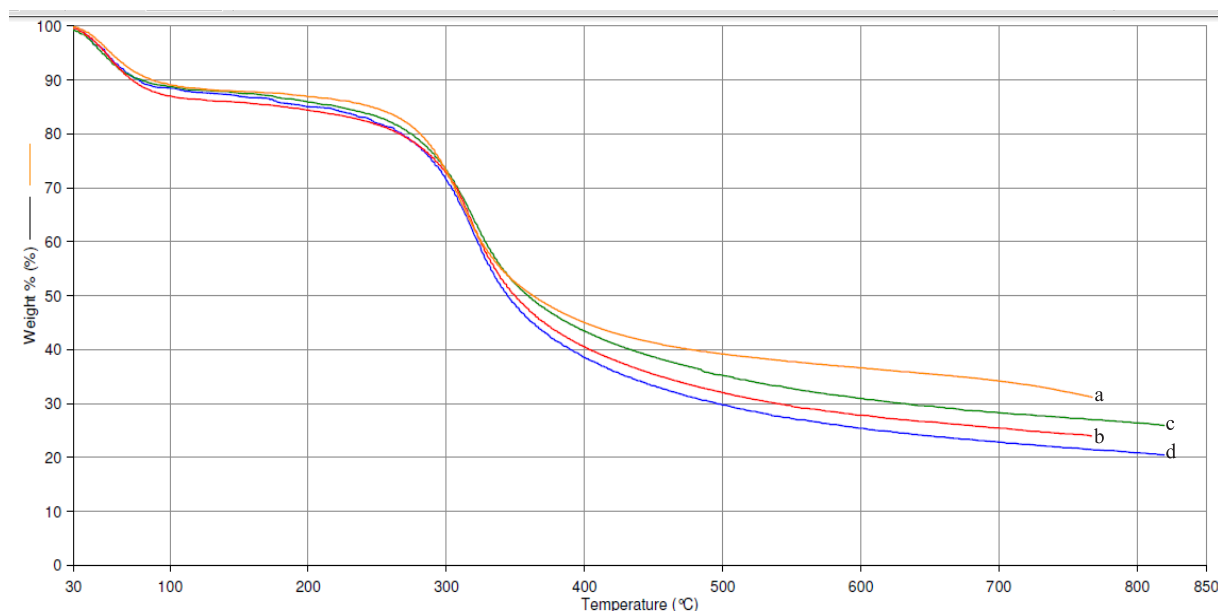


Figure 3. TG curves of (a) pickled goat skin, (b) vegetable tanned leather, (c) leather@SiO₂ and (d) leather@SiO₂/Ag

Based on the TG analysis, it was confirmed that the weight loss ratio was almost similar for all the samples. The first stage required a higher temperature than the other stages, corresponding

to the decomposition temperature of water. The slight differences in temperature of the second stage were identified for all samples.

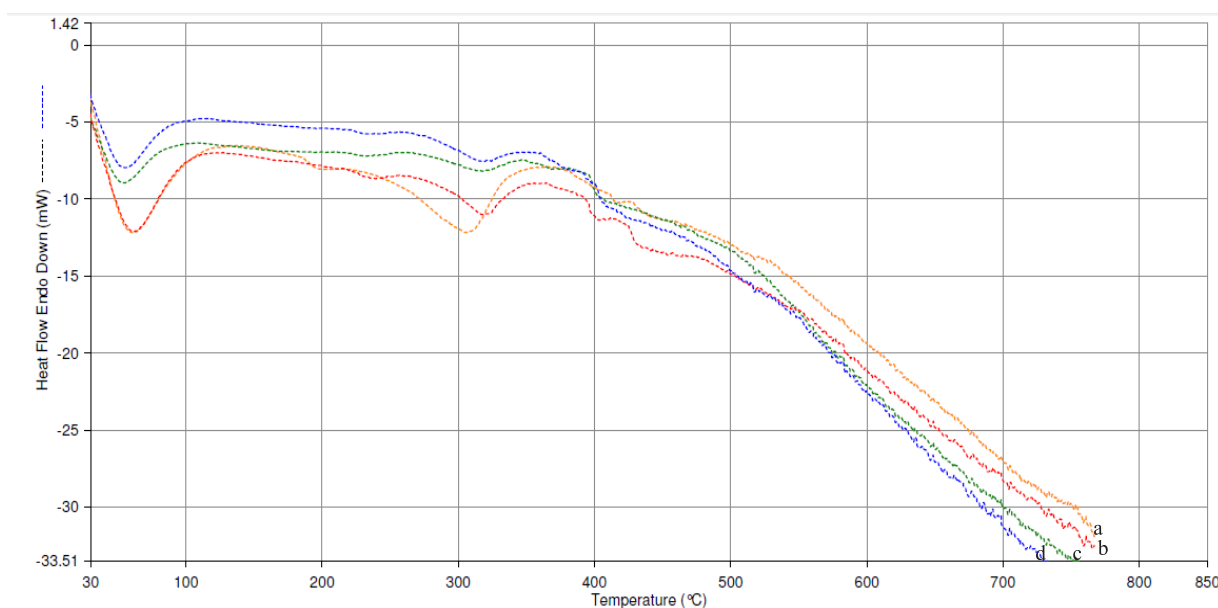


Figure 4. DTA curves of (a) pickled goat skin, (b) vegetable tanned leather, (c) leather@SiO₂ and (d) leather@SiO₂/Ag

The DTA curves informed two endothermic peaks of samples (Figure 4). The phase change observed at around 56 and 62°C corresponds to evaporation of water in pickled skin and coated leathers. The water evaporation temperatures of leather@SiO₂ and that of leather@SiO₂/

Ag were identified slightly lower than that of pickled goat skin and vegetable tanned leather. The increase of hydroxyl groups in leather@SiO₂ and leather@SiO₂/Ag can explain this difference in temperature. The endotherm at 300-325°C reflects to the release of organic moiety of

samples. The heat flow of pickled skin and the tanned leather were higher than the silica

treated leather, expressing that the coating of silica on leather increases the thermal stability of leather.

Antibacterial Activity

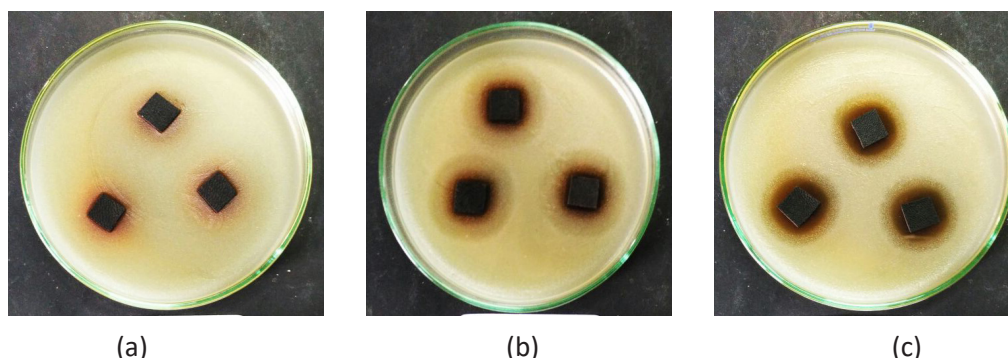


Figure 5. *Staphylococcus aureus* inhibition zone of (a) vegetable tanned leather, (b) leather@SiO₂, and (c) leather@SiO₂/Ag

The antibacterial activity assay was investigated by the agar diffusion method using *Staphylococcus aureus* to identify the bacteria inhibition zone diameter (Figure 5). The measurement data are written in Table 3. The results imply that the inhibition zones of leather@SiO₂ and leather@SiO₂/Ag were 21.25 ± 0.50 mm and 24.80 ± 1.64 mm, respectively. This area was larger than the inhibition zone of vegetable tanned leather that is equal to 11.40 ± 0.55 mm. Anova test shows a significant difference for every data. This confirmed the successful working of silica-based volcanic ash and silver as an antibacterial agent on vegetable tanned leather.

Table 3: Inhibition zone of samples

Sample	Inhibition Zone (mm)
Vegetable tanned leather	11.40 ± 0.55
Leather@SiO ₂	21.25 ± 0.50
Leather@SiO ₂ /Ag	24.80 ± 1.64

The antibacterial mechanism of silica is different from that of silver. Silica increased the hydrophobicity of vegetable tanned leather, thus reducing bacteria growth on vegetable tanned leather. While silver ions interact with disulfide or sulfhydryl groups of enzymes, causing structural changes leading to disruption of metabolic processes followed by cell death [35, 36]. Recently, it has been suggested that the

antibacterial mechanism of silver nanoparticles may also be related to membrane damage due to free radicals that are derived from the surface of the nanoparticles [37]. This bacterial activity also appeared to be dependent on the size and shape of the silver nanoparticles [38].

CONCLUSIONS

The utilization of silica-based volcanic ash modified with silver as an antibacterial agent in leather production was investigated in this study. The leather was obtained from the tanning process of goat skin using mimosa as a tanning agent. The antibacterial assay showed that the inhibition zones of *Staphylococcus aureus* for vegetable tanned leather, leather@SiO₂ and leather@SiO₂/Ag are 11.40 ± 0.55 mm, 21.25 ± 0.50 mm and 24.80 ± 1.64 mm, respectively. It verified the effectiveness of leather coating with silica-based volcanic ash/silver as an antibacterial agent.

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